ERRATUM for Energy and Entropy: A Dynamic Duo (Ed. 1)

Where?	Correction
p. 51, Key Point 2.13	The word "measurement" is incorrectly hyphenated.
p. 76, line 6	$d\zeta_1 - d\zeta_2$ should be $d\zeta_1 \neq -d\zeta_2$
p. 96, Eq. (4.1)	Right side should be $Nk\left[\frac{3}{2}\ln\left(\frac{E}{N}\right) + ln\left(\frac{V}{N}\right) + s_o\right]$ It is straightforward to show that S = Nk[ln(dimensionless quantity) + 5/2].
p. 96, 2 lines below Eq. (4.3)	Delete redundant word "at."
p. 98, par. 2, line 7	$\lvert\psi(q)\rvert^2\, dx$ should be $\lvert\psi(q)\rvert^2\, d^3q$
p. 102, Eq. (4.15)	Denominator should read $(\exp(\epsilon - \mu)/kT)$
p. 109, Eq. (4.26)	The term $C_v\, dT$ is missing in the equation's last line. Replace the words leading to Eq. (4.27) by: The internal energy can be found by integrating the corrected third line with respect to *V* along a path of constant *T,* giving $E(T,V) = -aN^2/V + \chi(T,a,b)$. The last term is a "constant" of integration along a path where *T* is constant. Equality of the mixed derivatives of *E(T, V),* along with the *T*-independence of aN^2/V guarantee that C_V is independent of *V.* For V→∞, $C_v \rightarrow 3/2$NkT, so for all *a, b* and *V,* $\chi(T,a,b) = 3/2\, NkT = \chi(T)$ and Eq. (4.27) is verified.
p. 110, par. 2	Assuming spherically symmetric interparticle forces and potential energies for a homogeneous and isotropic fluid, the probability of finding a second molecule in a spherical shell of thickness *dr* at distance *r* from a given molecule is $g(r)dV = g(r)\, 4\pi\, r^2\, dr$.
p. 140, last line	Delete "in" following T_2
p. 157, par. 1	Volume should be $V = 3\,m \times 4\,m \times 5\,m = 60\,m^3$
p. 296, before last paragraph.	Just as temperature is a measure of *hotness*, the Caratheodory development provides an interpretation of entropy as a measure of the adiabatic accessibility of states: State *B* is reversibly accessible from state *A* if $S_B = S_A$, is irreversibly accessible from *A* if $S_B > S_A$, and is inaccessible from A if $S_B < S_A$. (This insight was kindly provided to me by Prof. Chris Gray.)

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